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Internal Webinar

Effect of aryl π -conjugation of photoluminescent amino and ammoniumboranes for selective sensing on picric acid and F-, CN-ions

Ramar Arumugam

TIFR, Hyderabad

Molecular based sensors for the detection of small ions (anions/cations) have been received attention in industrial and medical field.[1] Sensors should have an excellent selectivity, stability, eco-friendly, water soluble low detection limit and more prone to response. The boron-ligand complexes are prominent to sense anions like fluoride and cyanide owing to their intrinsic characteristic features such as Lewis acidity and tunability. Using this fascinating property, a large number of organoboron systems have been constructed for various applications such as sensing, bio-probes, and optical devices.[1] In view of this interest we synthesized a series ammonium boranes and neutral amino boranes from different bromo-N,N-dimethylamine precursors. These compounds are airand moisture-stable and soluble in organic solvents. These compounds were characterized by multinuclear NMR spectroscopic methods as well as single-crystal X-ray diffraction analysis. The photophysical properties of these compounds are being investigated.

References:

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